
14TH Multidimensional Chromatography Workshop



Quantitative Analysis of Aliphatic Olefins in Fuels made from Plastic Waste by Comprehensive Two-Dimensional Gas Chromatography

Petr Vozka, Ph.D.

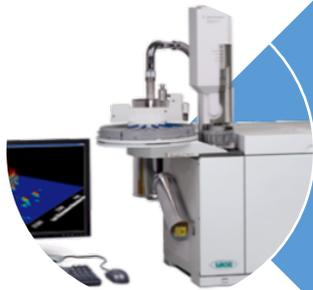
Assistant Professor

Chemistry & Biochemistry

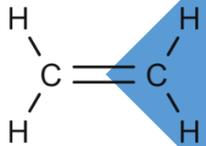
California State University, Los Angeles

January 30, 2023





Comprehensive Two-Dimensional Gas Chromatography (GC×GC)

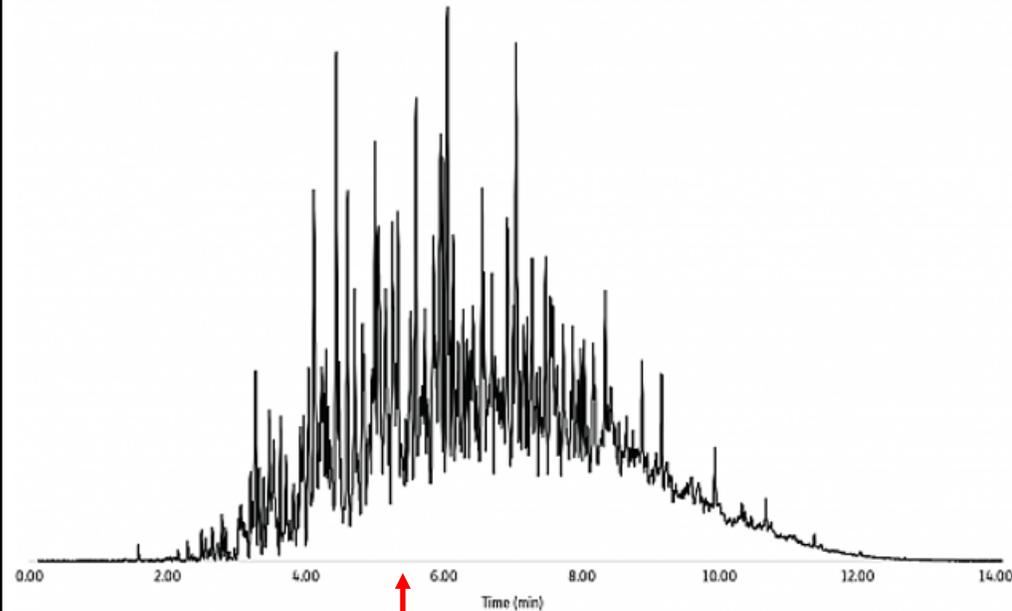


Quantitative Analysis of Aliphatic Olefins in Fuels made from Plastic Waste by GC×GC

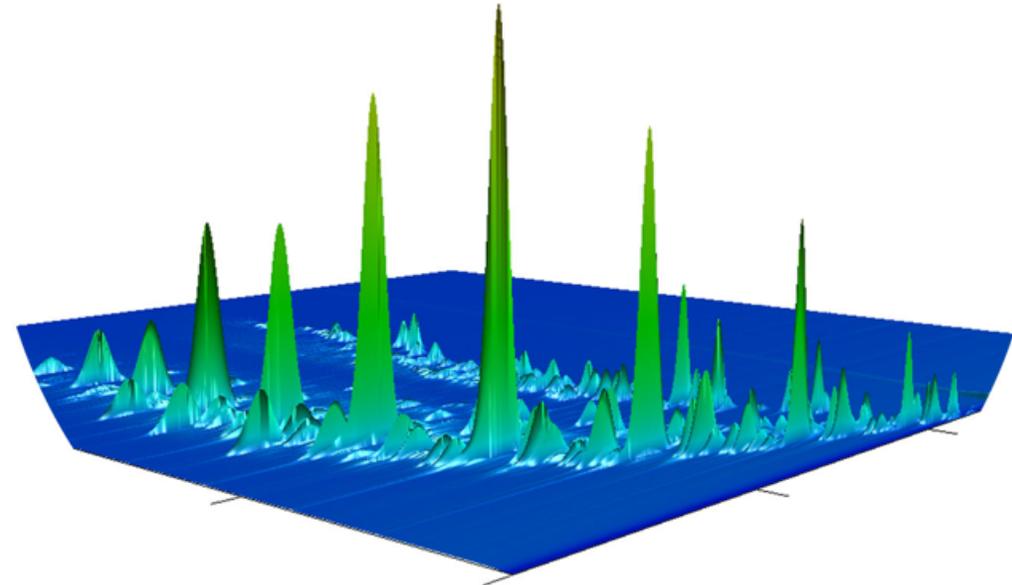


GC×GC in Academia/Teaching

How it started:



How it's going:



peak capacity limited!

GCxGC Chromatogram

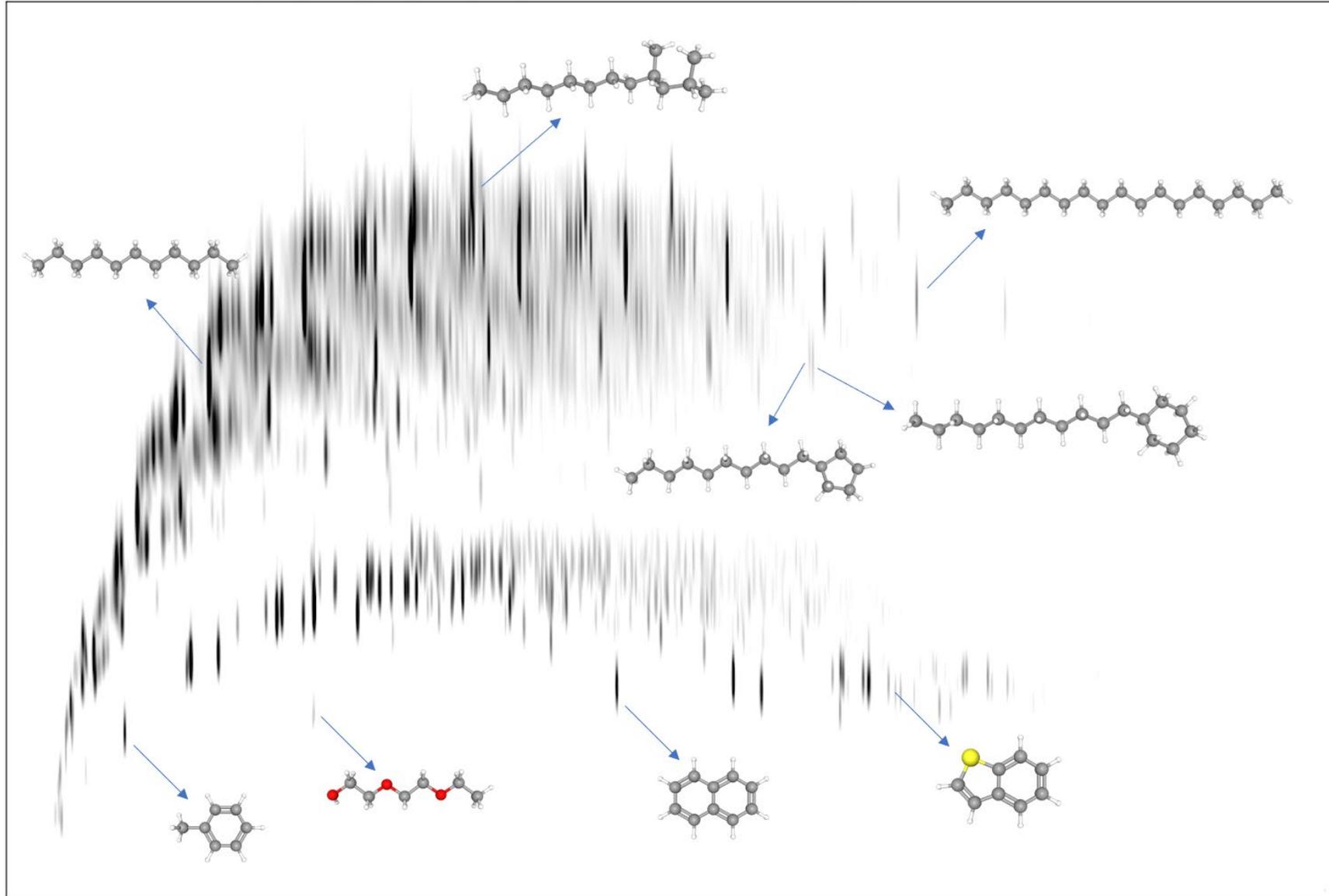
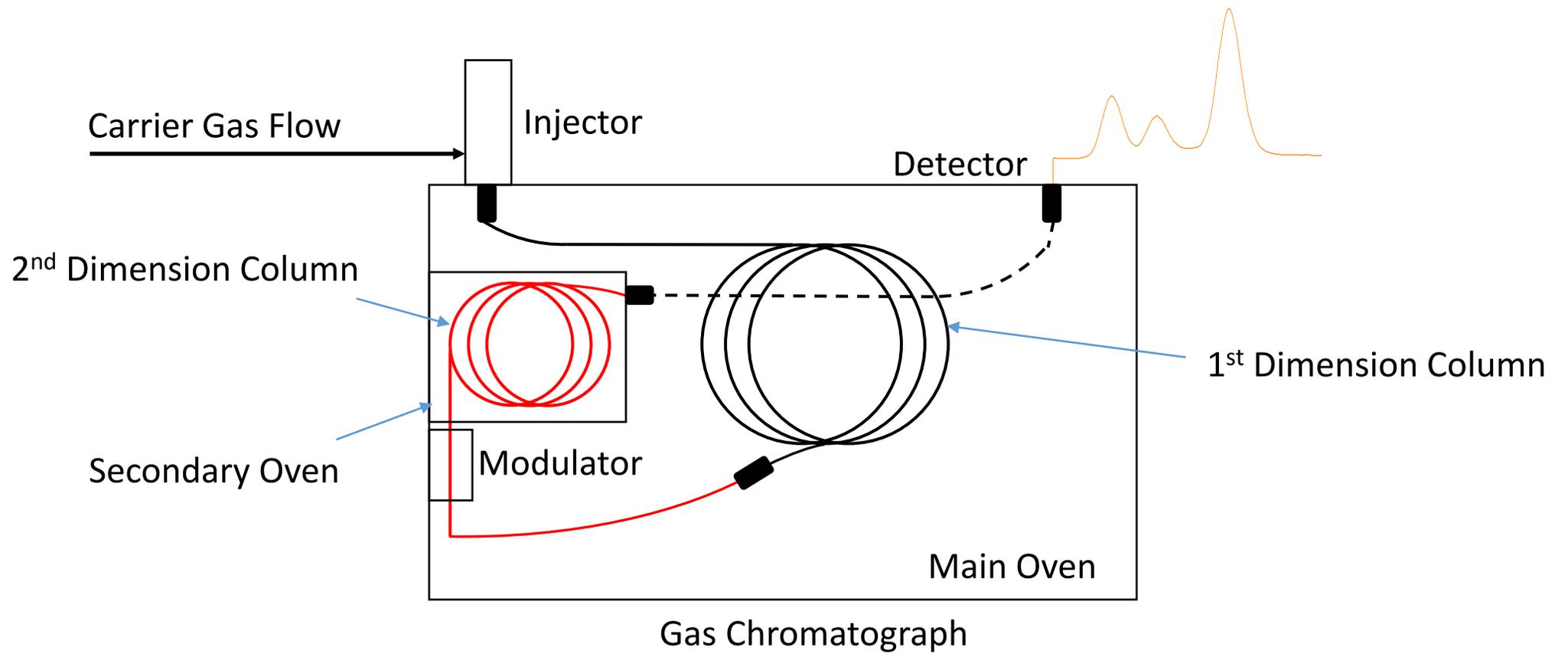
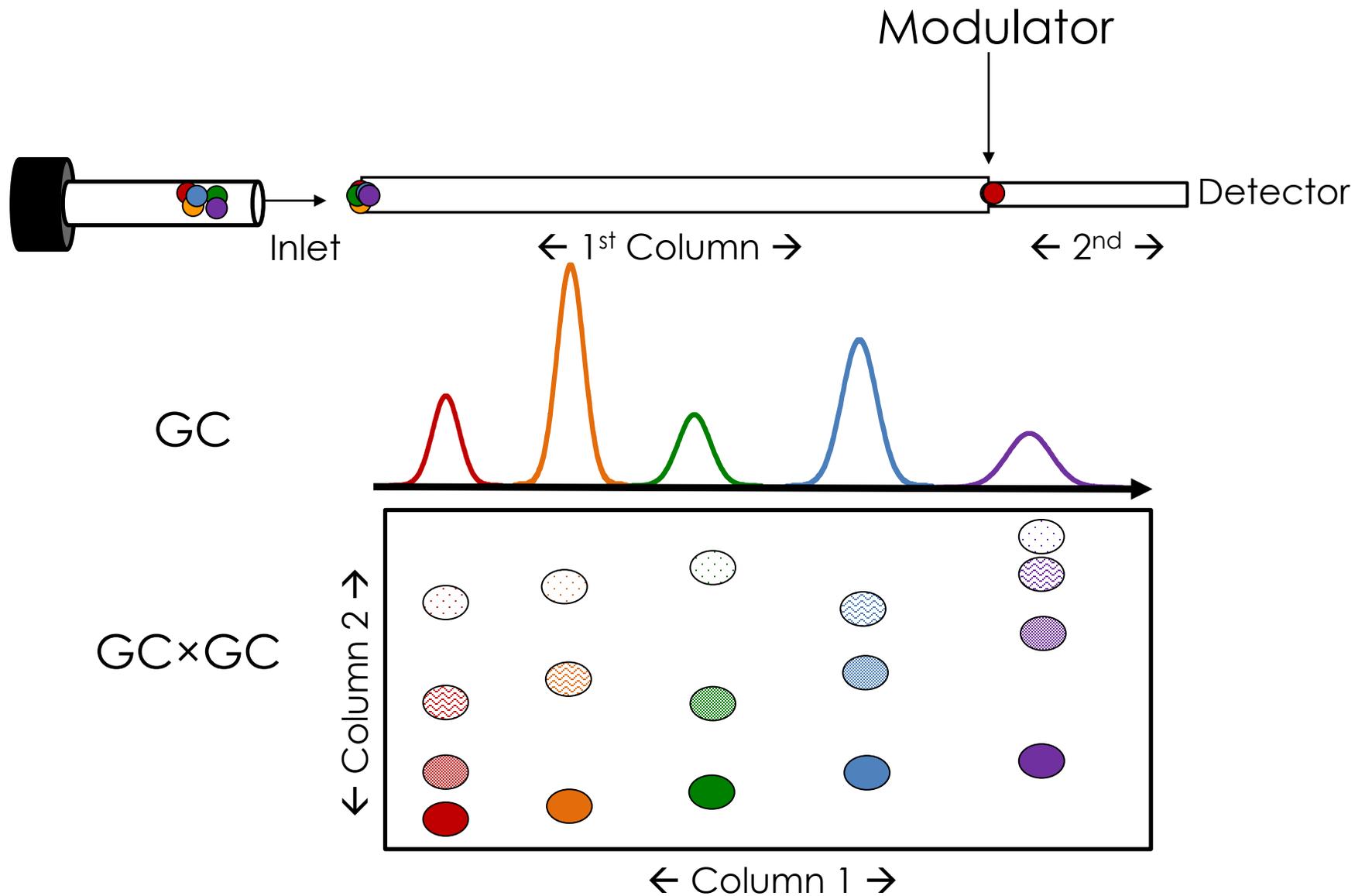


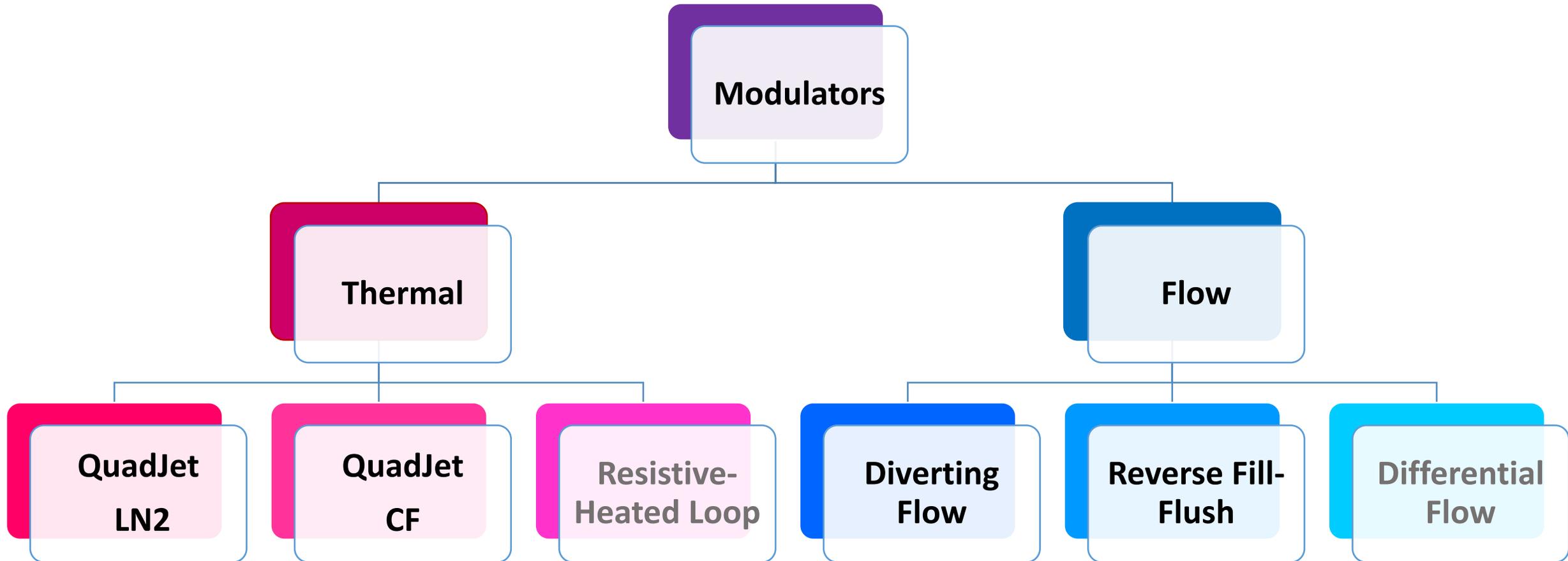
Diagram of a GC×GC Instrument



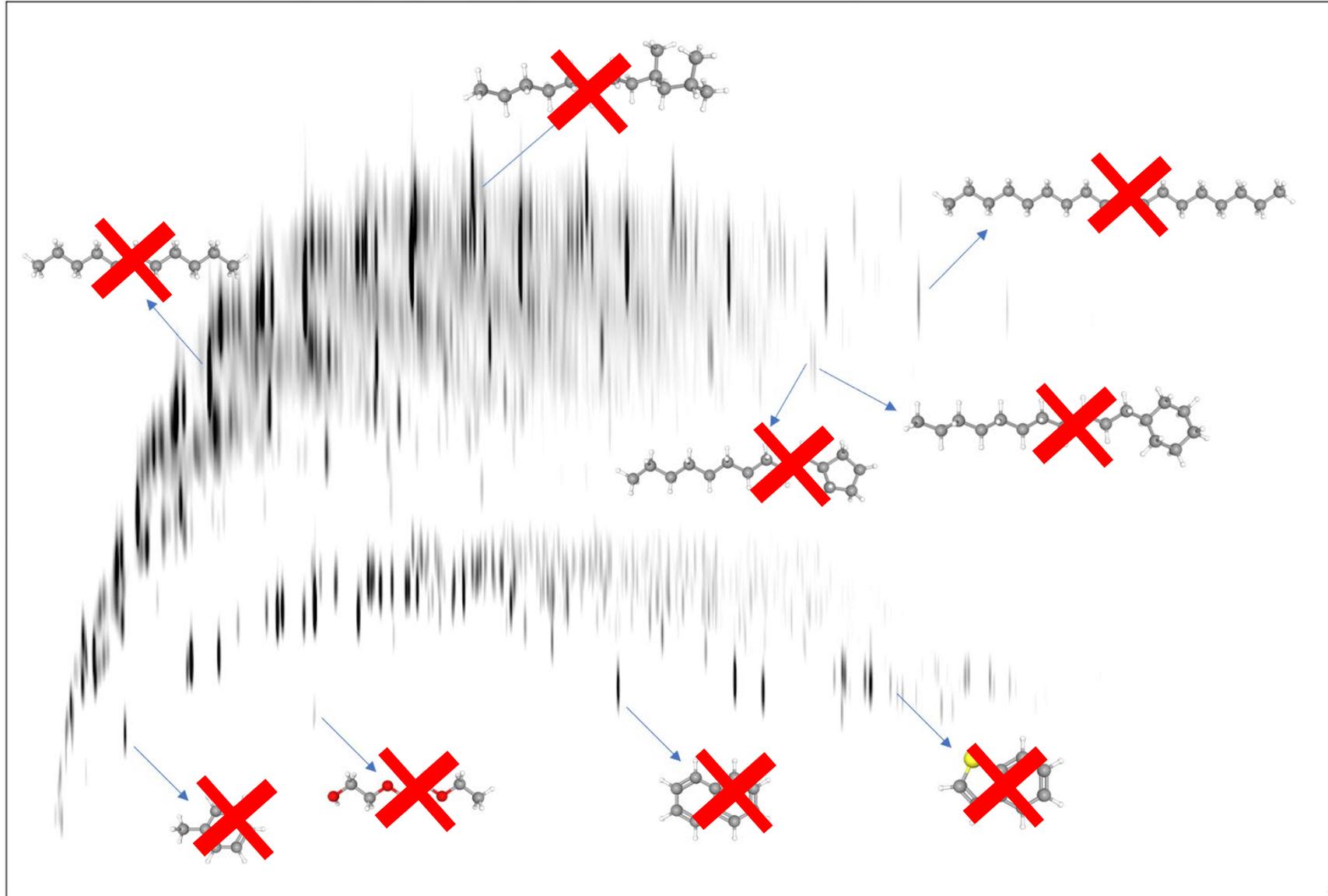
How GC×GC is achieved

1. Sample is injected on set of two columns connected in series
2. Primary column separates analytes in typical GC way
3. Analytes are modulated and then released onto secondary column
4. Secondary column separates analytes further with complementary phase chemistry

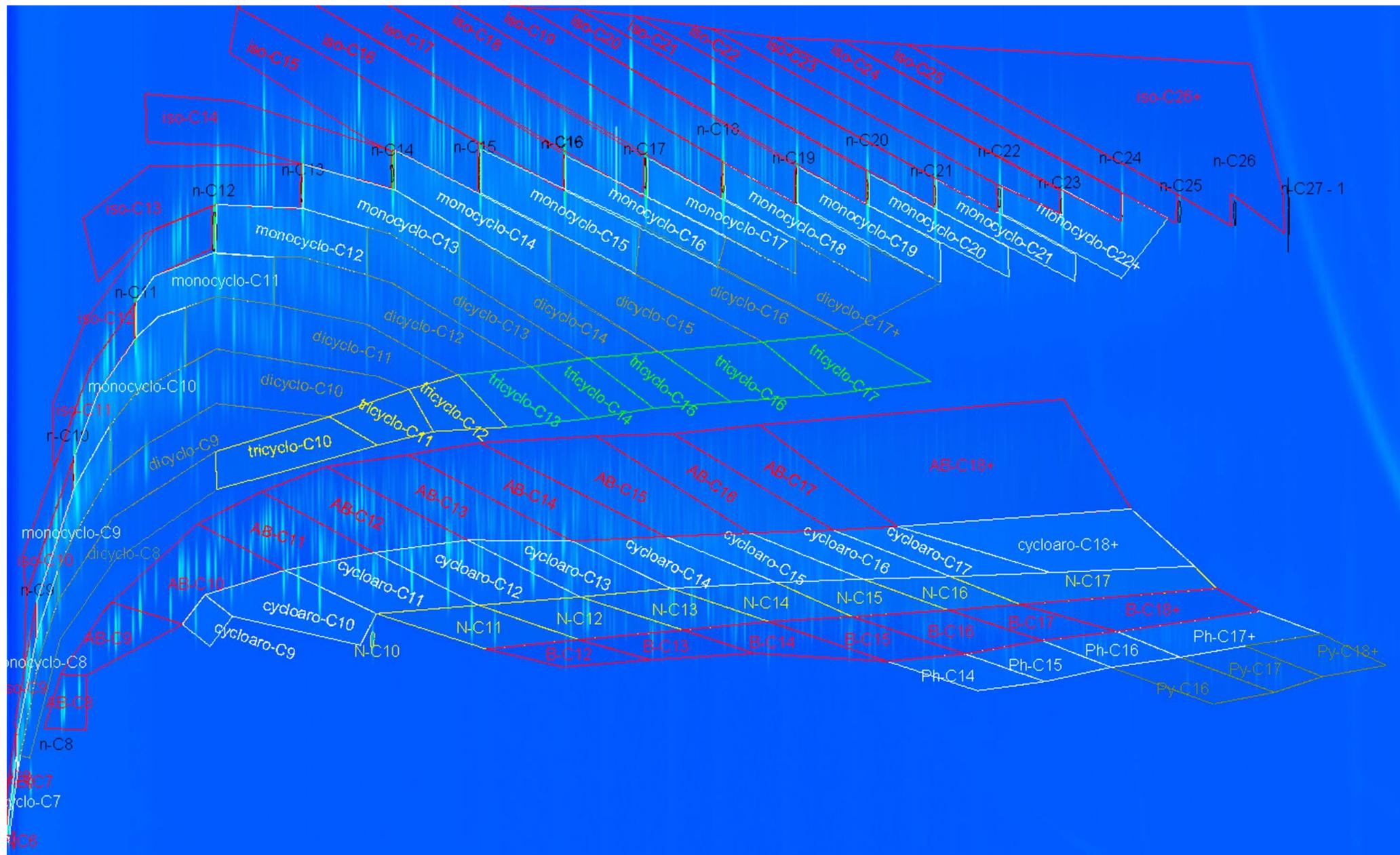




GCxGC Chromatogram



GCxGC Classification



Quantitative results (some)



n-paraffins	F-76	Jet A	FT-IPK	Green diesel
C8	0.13	0.83	0.00	0.13
C9	0.42	5.05	0.00	0.20
C10	1.54	4.96	0.10	0.18
C11	2.32	3.36	0.00	0.00
C12	2.22	2.37	0.10	0.18
C13	2.21	1.90	0.08	0.23
C14	2.13	1.27	0.04	0.40
C15	1.93	0.76	0.03	0.88
C16	1.71	0.36	0.01	2.84
C17	1.58	0.10	0.00	1.76
C18	1.32	0.02	0.00	4.40
C19	1.10	0.00	0.00	0.04
C20	0.95	0.00	0.00	0.08
C21	0.72	0.00	0.00	0.00
C22	0.45	0.00	0.00	0.01
C23	0.24	0.00	0.00	0.00
C24	0.11	0.00	0.00	0.00
C25	0.05	0.00	0.00	0.00
C26	0.02	0.00	0.00	0.00
C27	0.00	0.00	0.00	0.00
Total n-paraffins	21.15	20.97	0.35	11.33

	F-76	Jet A	FT-IPK	Green diesel
Alkylbenzenes				
C7	0.06	0.07	0.00	0.03
C8	0.26	1.79	0.01	0.00
C9	1.30	4.86	0.07	0.00
C10	1.75	3.27	0.08	0.00
C11	1.33	2.15	0.04	0.00
C12	0.94	1.72	0.00	0.00
C13	0.63	1.04	0.00	0.00
C14	0.33	0.35	0.00	0.00
C15	0.25	0.19	0.00	0.00
C16	0.20	0.02	0.00	0.00
C17	0.19	0.00	0.00	0.00
C18 +	0.14	0.00	0.00	0.00
Total alkylbenzenes	7.40	15.46	0.20	0.03
Cycloaromatics				
C9	0.05	0.14	0.00	0.00
C10	0.44	0.78	0.00	0.00
C11	1.29	1.73	0.01	0.00
C12	1.68	2.24	0.05	0.00
C13	1.52	1.26	0.01	0.00
C14	1.19	0.73	0.00	0.00
C15	1.02	0.01	0.00	0.00
C16	0.36	0.00	0.00	0.00
C17	0.03	0.00	0.00	0.00
C18 +	0.00	0.00	0.00	0.00
Total Cycloaromatics	7.58	6.89	0.08	0.00

Detailed Analysis of Complex Chemical Mixtures



Such as:

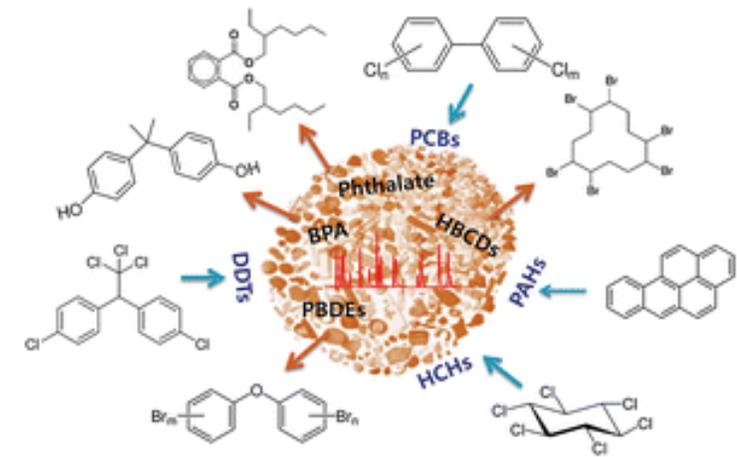
- Petroleum products



- Beach Oil spills



- Microplastics analysis
(Organic compounds)



Plastic Waste in Oceans



Ocean Of Plastic



© picture-alliance/Photoshot



Plastic Waste in Oceans





Experts say that by 2050 there may be more plastic than fish in the ocean, or perhaps only plastic left.

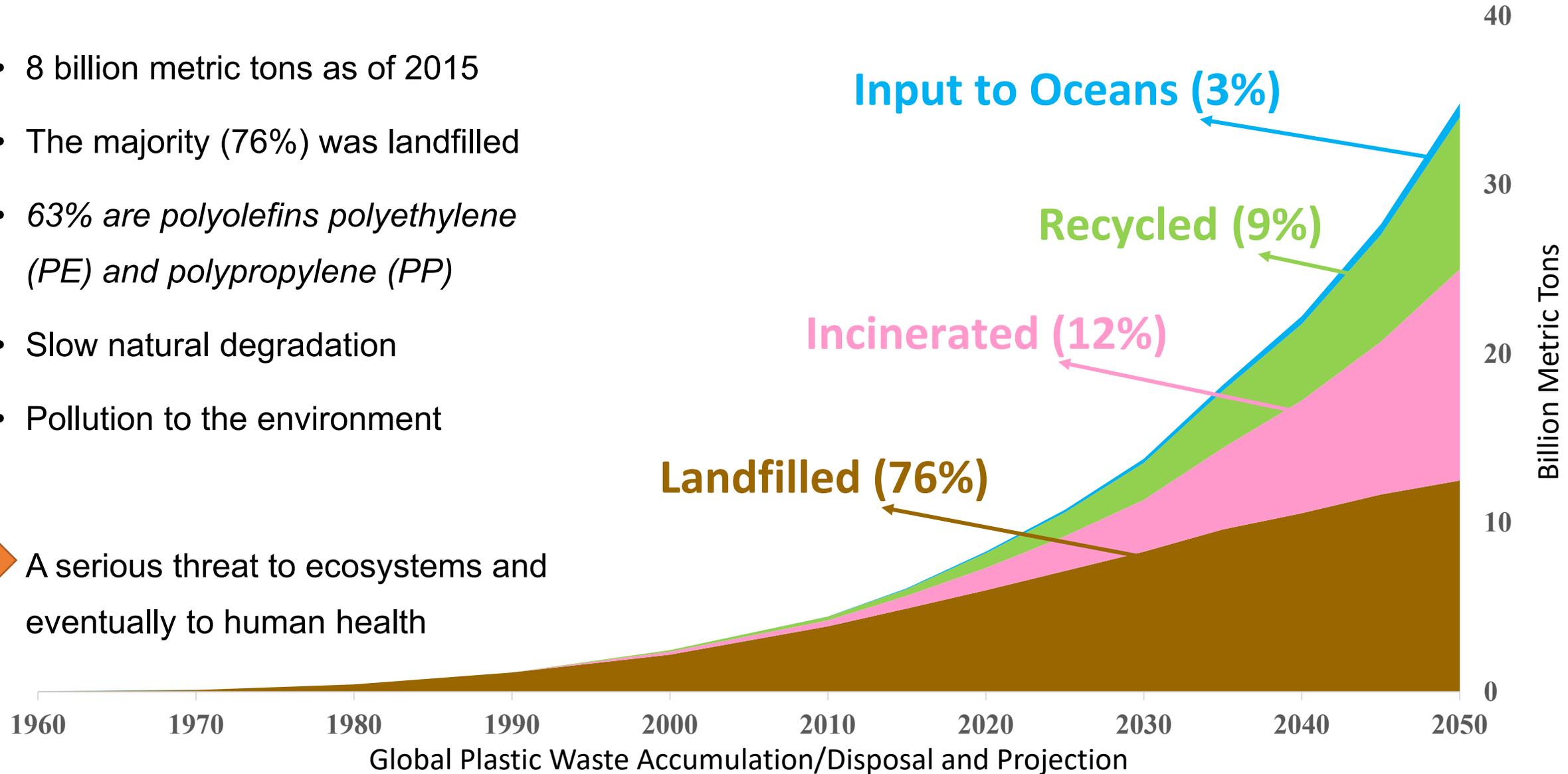
Plastic Waste - Landfill



The growing plastic waste problem

- 8 billion metric tons as of 2015
- The majority (76%) was landfilled
- 63% are polyolefins polyethylene (PE) and polypropylene (PP)
- Slow natural degradation
- Pollution to the environment

➔ A serious threat to ecosystems and eventually to human health



Chemical conversion of plastic waste into fuels



UNIVERSITY OF
CHEMISTRY AND TECHNOLOGY
PRAGUE

- Pyrolysis (+ hydrotreating) of:
plastic foils and waste tires



- Hydrothermal Processing of:
polyolefin plastic waste



Fuel 273 (2020) 117726



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Fuel

journal homepage: www.elsevier.com/locate/fuel



Full Length Article

Conversion of polyethylene waste into clean fuels and waxes via hydrothermal processing (HTP)

Kai Jin^{a,b,1}, Petr Vozka^{b,1}, Gozdem Kilaz^b, Wan-Ting Chen^c, Nien-Hwa Linda Wang^{a,*}

^a Davidson School of Chemical Engineering, Purdue University, West Lafayette, IN, 47907, USA

^b School of Engineering Technology, Fuel Laboratory of Renewable Energy (FLORE), Purdue University, West Lafayette, IN

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Fuel 294 (2021) 120505

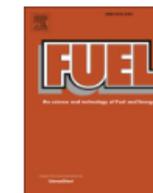


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Full Length Article

Low-pressure hydrothermal processing of mixed polyolefin wastes into clean fuels

Kai Jin^a, Petr Vozka^b, Clayton Gentilcore^c, Gozdem Kilaz^a, Nien-Hwa Linda Wang^{c,*}

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Production of Transportation Fuels Via Hydrotreating of Scrap Tires Pyrolysis Oil

22 Pages • Posted: 1 Nov 2022

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Under Review



Chemical Engineering Journal

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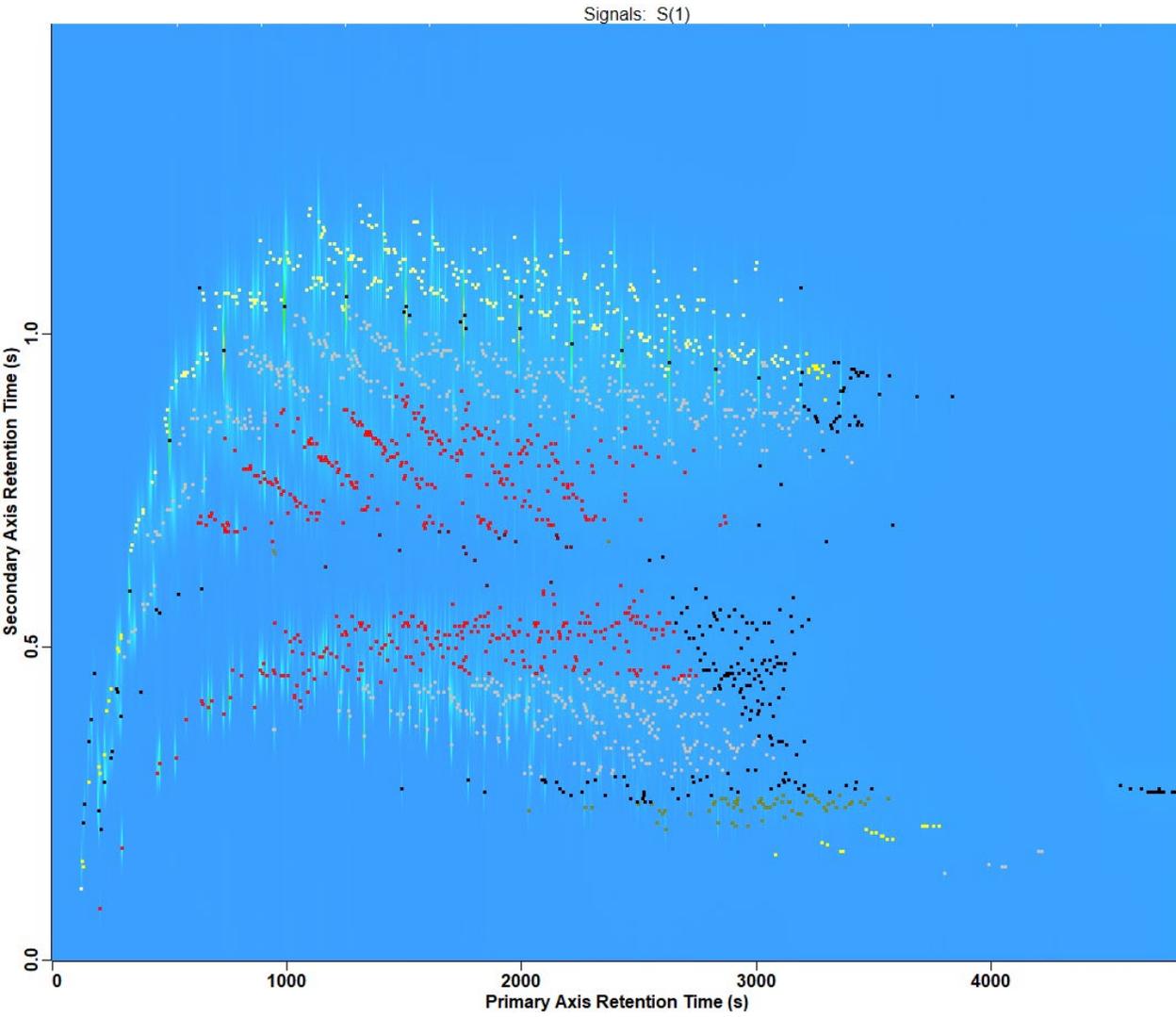
19.4

CiteScore

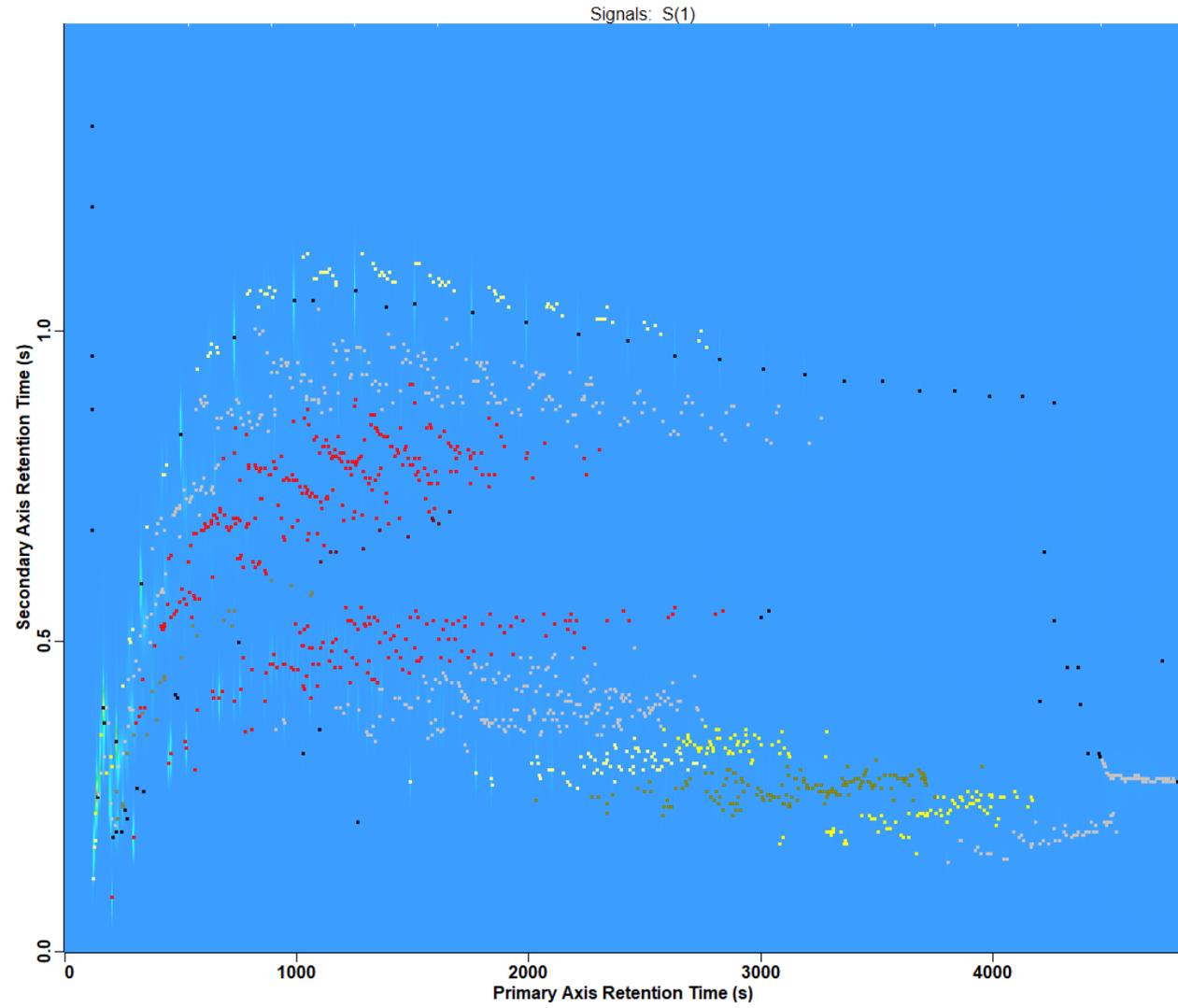
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Impact Factor

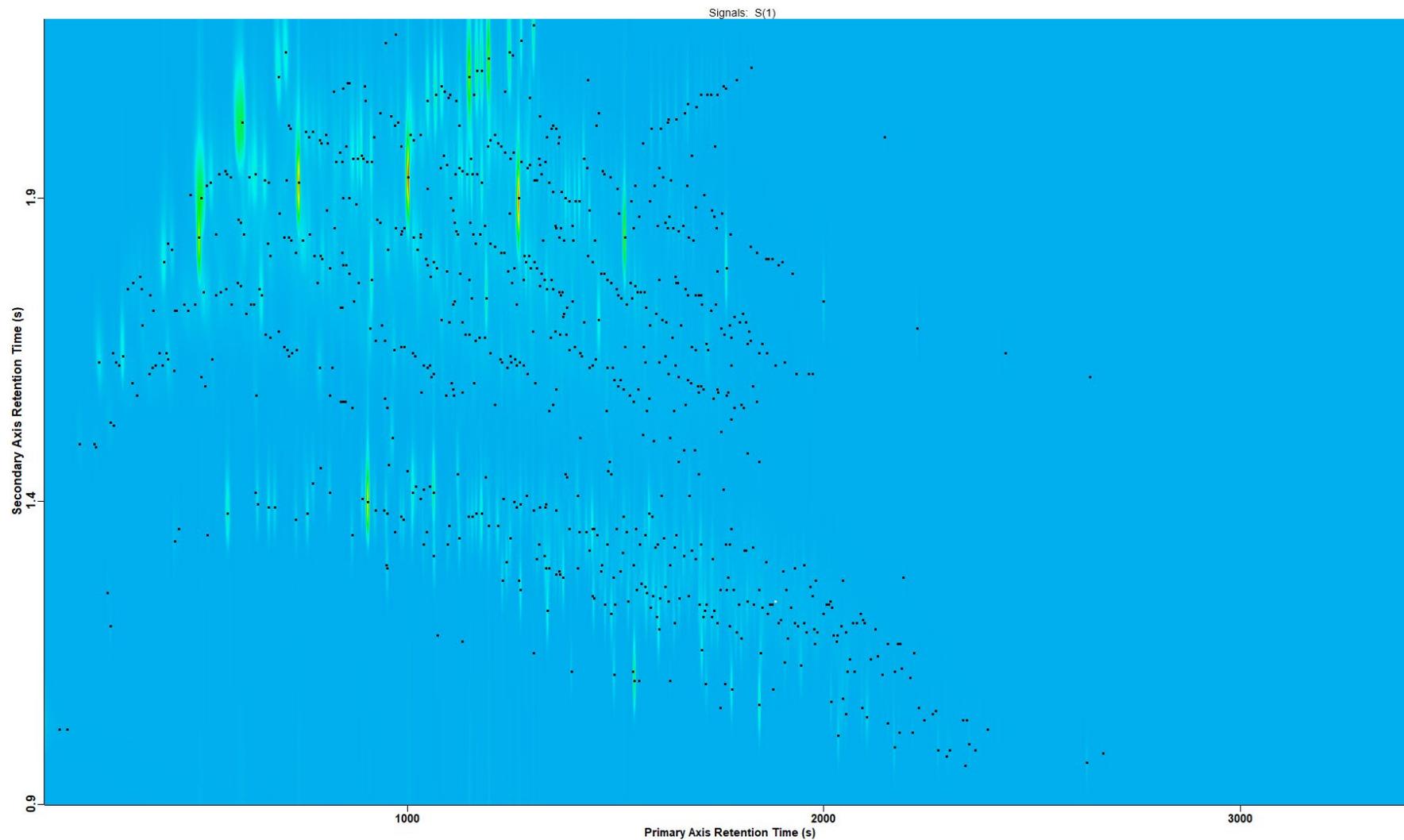
Commercial Diesel fuel

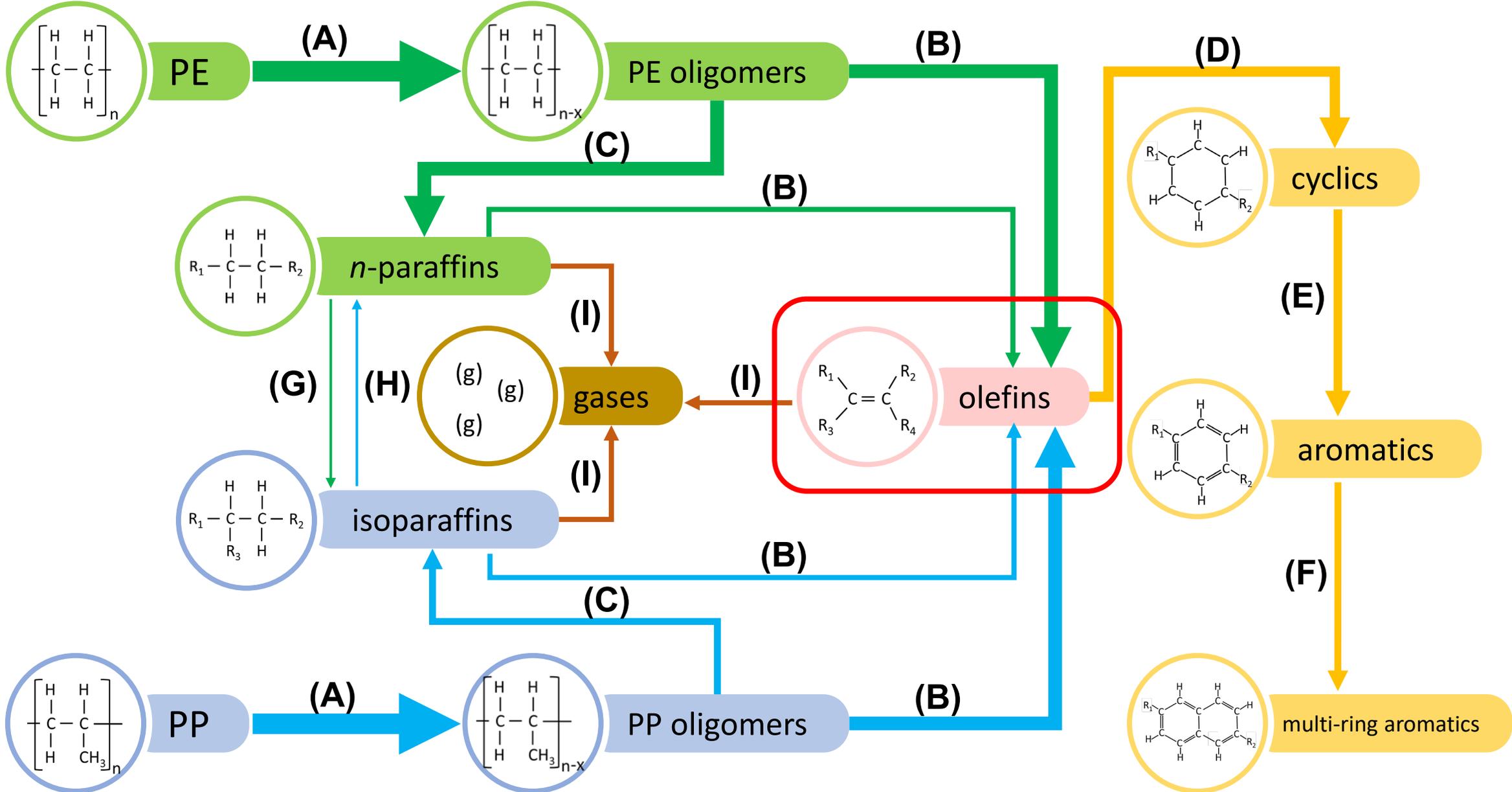


HTP Diesel fuel

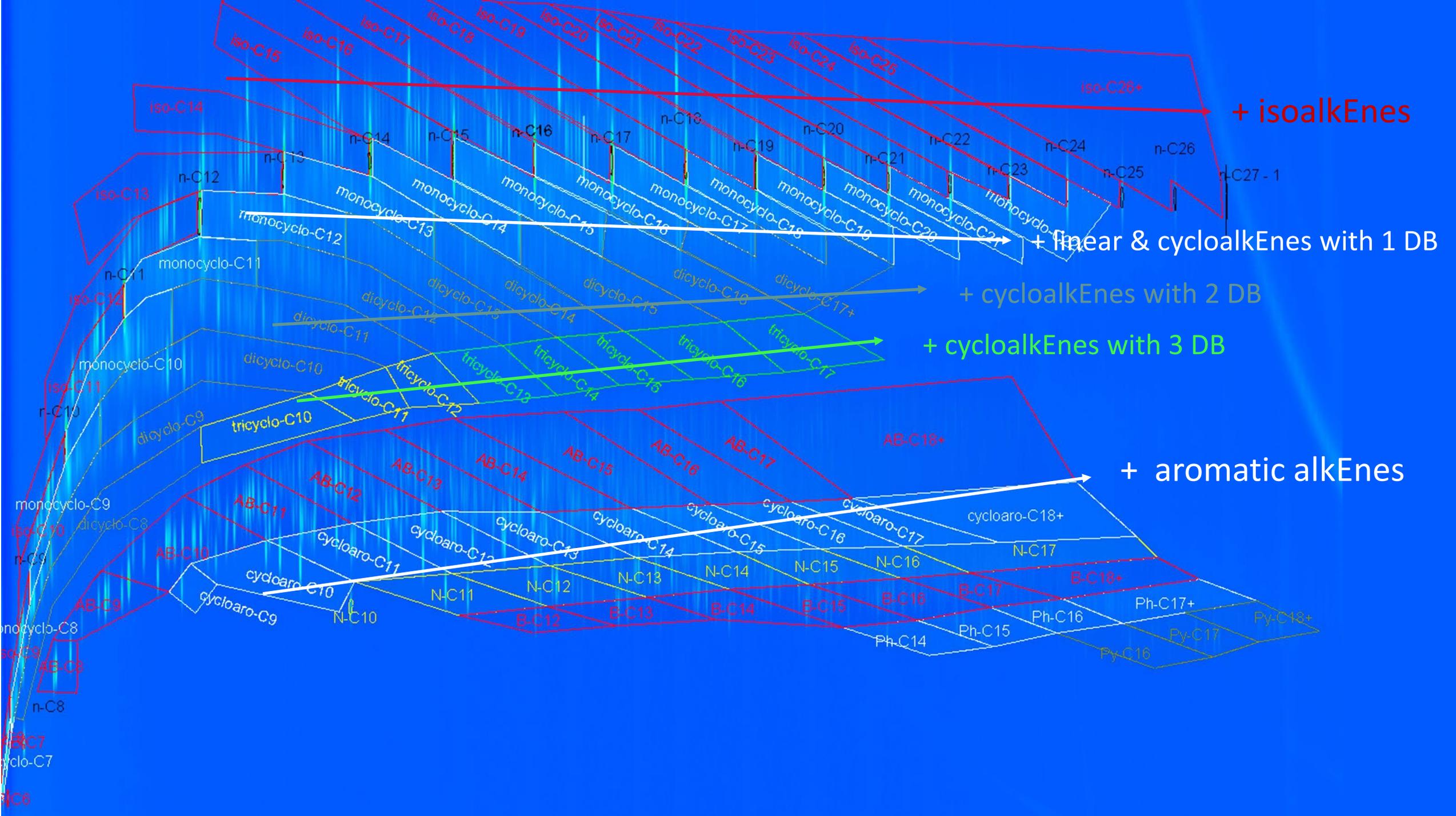


Diesel fuel distillation range pyrolysis oil after hydrotreating (270 °C and 6 MPa)





Potential reaction pathways of PE and PP co-processing under LP-HTP. (A) depolymerization, (B) β -scission, (C) hydrogen abstraction, (D) cyclization, (E) dehydrogenation, (F) formation of multi-ring aromatics, (G) isomerization, (H) formation of short *n*-paraffins (C_{6-7}), (I) further cracking to gases. The thickness of the arrows indicates the relative amounts of products.



+ isoalkenes

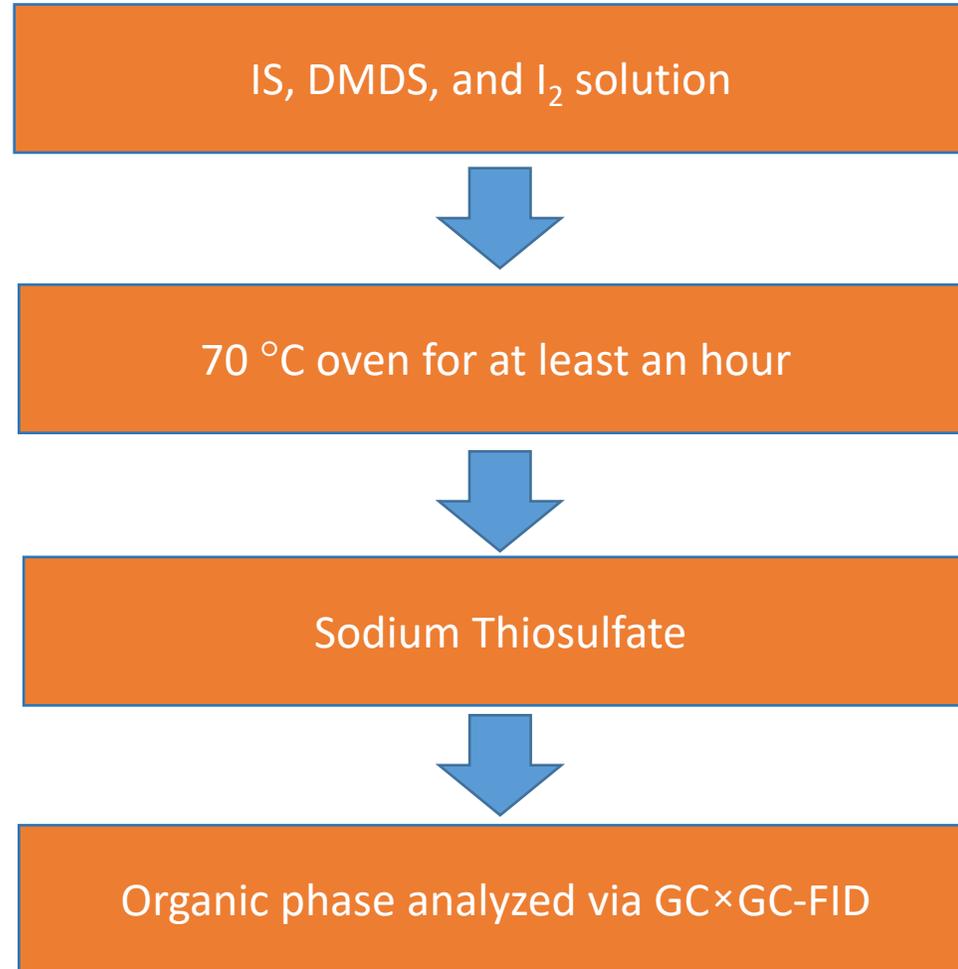
+ linear & cycloalkenes with 1 DB

+ cycloalkenes with 2 DB

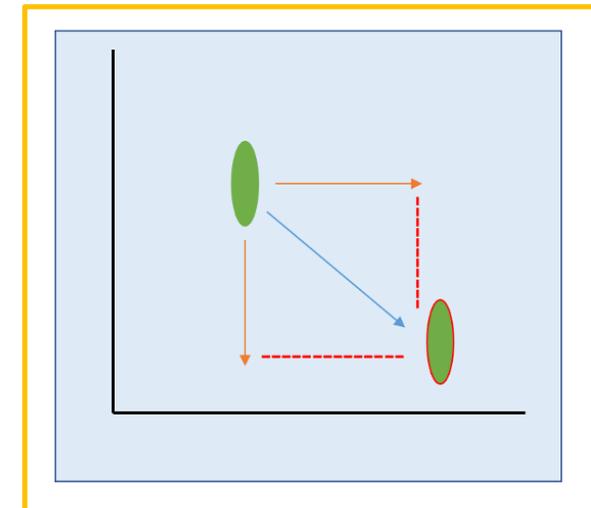
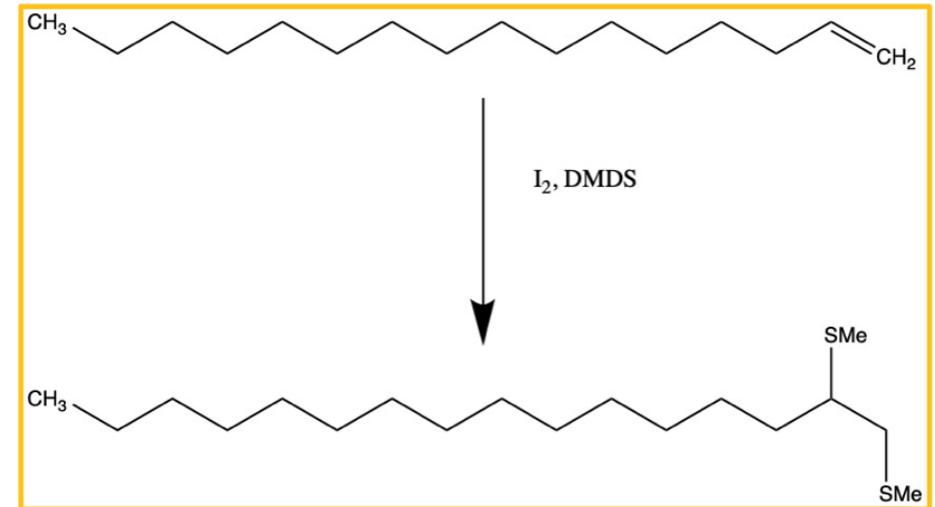
+ cycloalkenes with 3 DB

+ aromatic alkenes

Sample Preparation



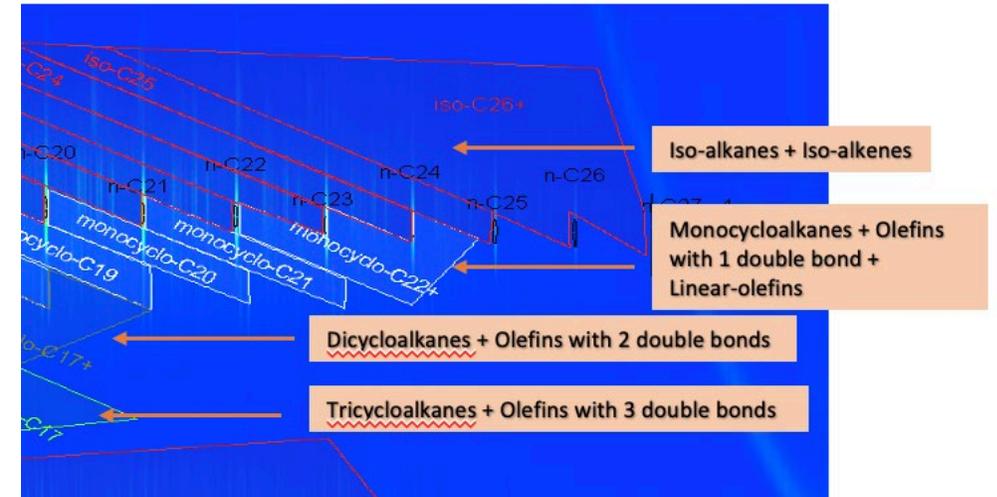
Theory



Calculations

Olefins in sample

- Iso-alkenes
- Olefins with 1 double bond + Linear-alkenes
- Olefins with 2 double bonds
- Olefins with 3 double bonds



Equation

$$Wt. \% Olefin, C\# = P.A. Pre-Derivatization, C\# - P.A. Post-Derivatization and Normalization, C\#$$

Example

$$Wt. \% Iso-alkene, C11 = P.A. Pre-Derivatization, C11 - P.A. Post-Derivatization and Normalization, C11$$

Calculations

Olefins in sample

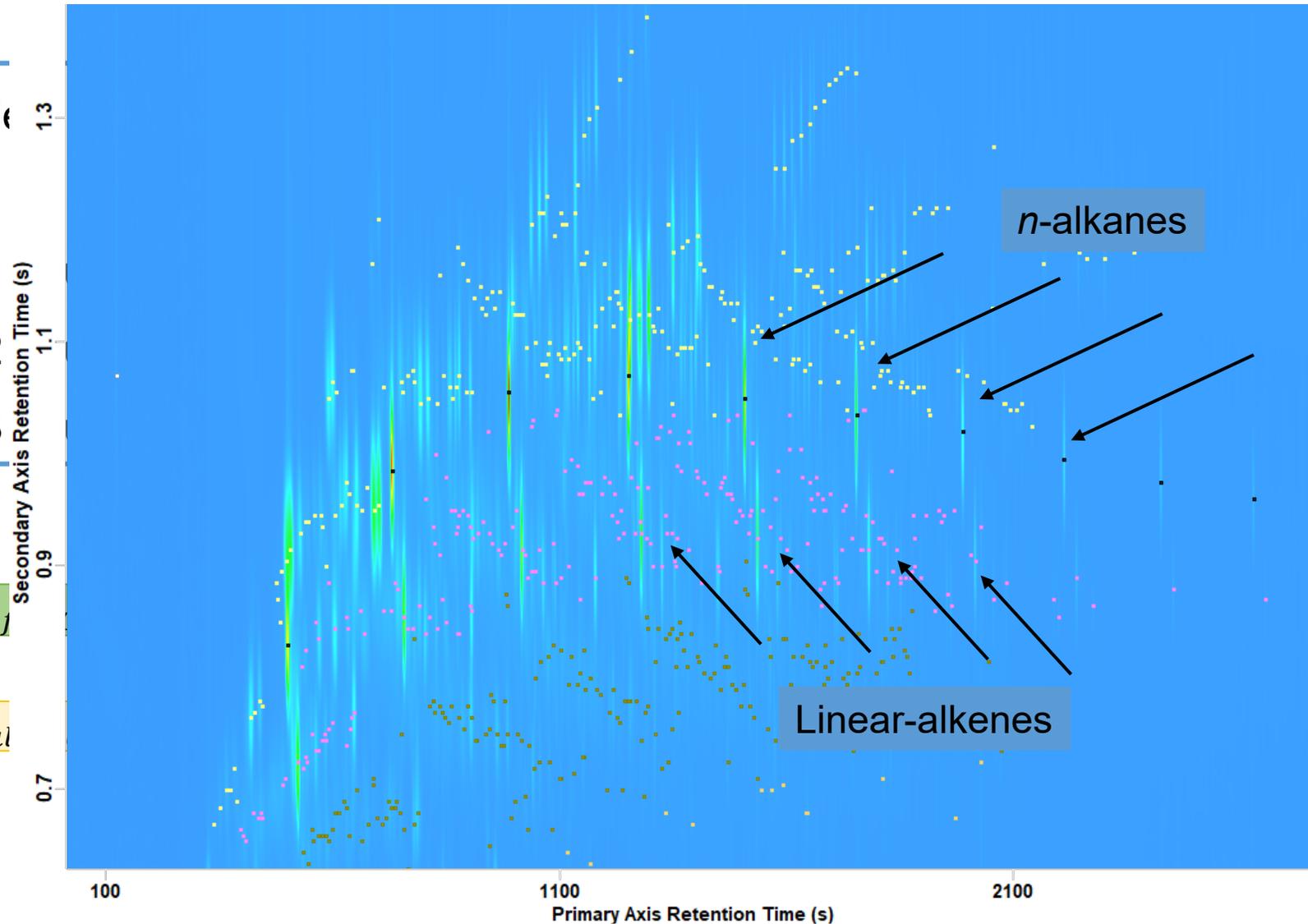
- Iso-alkenes
- Olefins with 1
- Olefins with 2
- Olefins with 3

Equation

$Wt. \% olef$

Example

$Wt. \% Iso-alk$

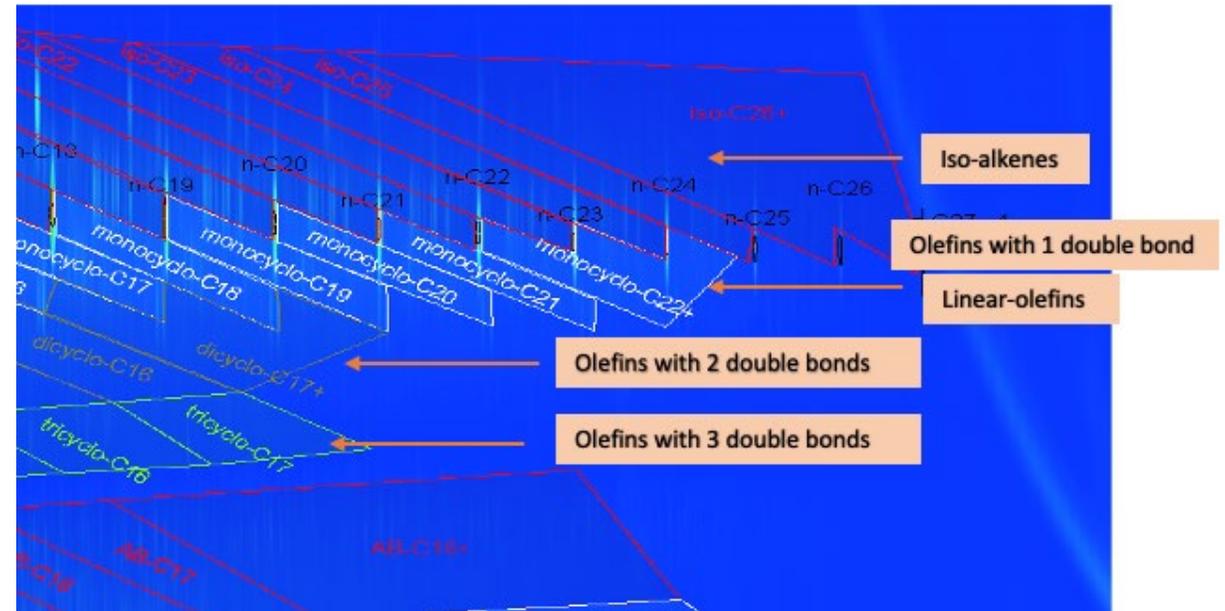


C11

Calculations

Olefins in sample

- Linear-alkenes
- **Iso-alkenes**
- Olefins with 1 double bond
- Olefins with 2 double bonds
- Olefins with 3 double bonds



Equation

$$Wt. \% Olefin, C\# = P.A. Pre-Derivatization, C\# - P.A. Post-Derivatization and Normalization, C\#$$

Example

$$Wt. \% Iso-alkene, C11 = P.A. Pre-Derivatization, C11 - P.A. Post-Derivatization and Normalization, C11$$

Example

$$Wt. \% \text{ monocycloalkene, } C_{11} = P.A. \text{ Pre-Derivatization, } C_{11} - P.A. \text{ Post-Derivatization and Normalization, } C_{11}$$

UCT Prague sample; diesel fraction obtained from the pyrolysis of scrap tires

Cycloalkanes	Surovina PNEU PE (area)	Surovina PNEU PE (wt. %)	Post-Der. (area)	Post-Der. Normalization (area)	Olefins (area)	Olefin (wt. %)	Real Cyclo (wt. %)
monocyclo-alkane C5	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C6	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C7	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C8	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C9	2457.75	0.46	525.59	417.13	2040.62	0.38	0.08
monocyclo-alkane C10	5073.85	0.94	4278.24	3395.43	1678.42	0.31	0.63
monocyclo-alkane C11	8531.58	1.59	6142.81	4875.25	3656.33	0.68	0.91
monocyclo-alkane C12	3023.63	0.56	1665.03	1321.45	1702.18	0.32	0.25
monocyclo-alkane C13	2498.03	0.46	1307.33	1037.56	1460.47	0.27	0.19
monocyclo-alkane C14	1658.46	0.31	510.81	405.40	1253.06	0.23	0.08
monocyclo-alkane C15	382.18	0.07	0	0	382.18	0.07	0.00
monocyclo-alkane C16	201.36	0.04	0	0	201.36	0.04	0.00
monocyclo-alkane C17	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C18	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C19	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C20	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C21	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C22	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C23	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C24	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C25	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C26	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C27	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C28	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C29	0	0.00	0	0	0	0.00	0.00
monocyclo-alkane C30+	0	0.00	0	0	0	0.00	0.00
total monocyclo-alkanes	23826.84	4.43	14429.81	11452.23	12374.61	2.30	2.13

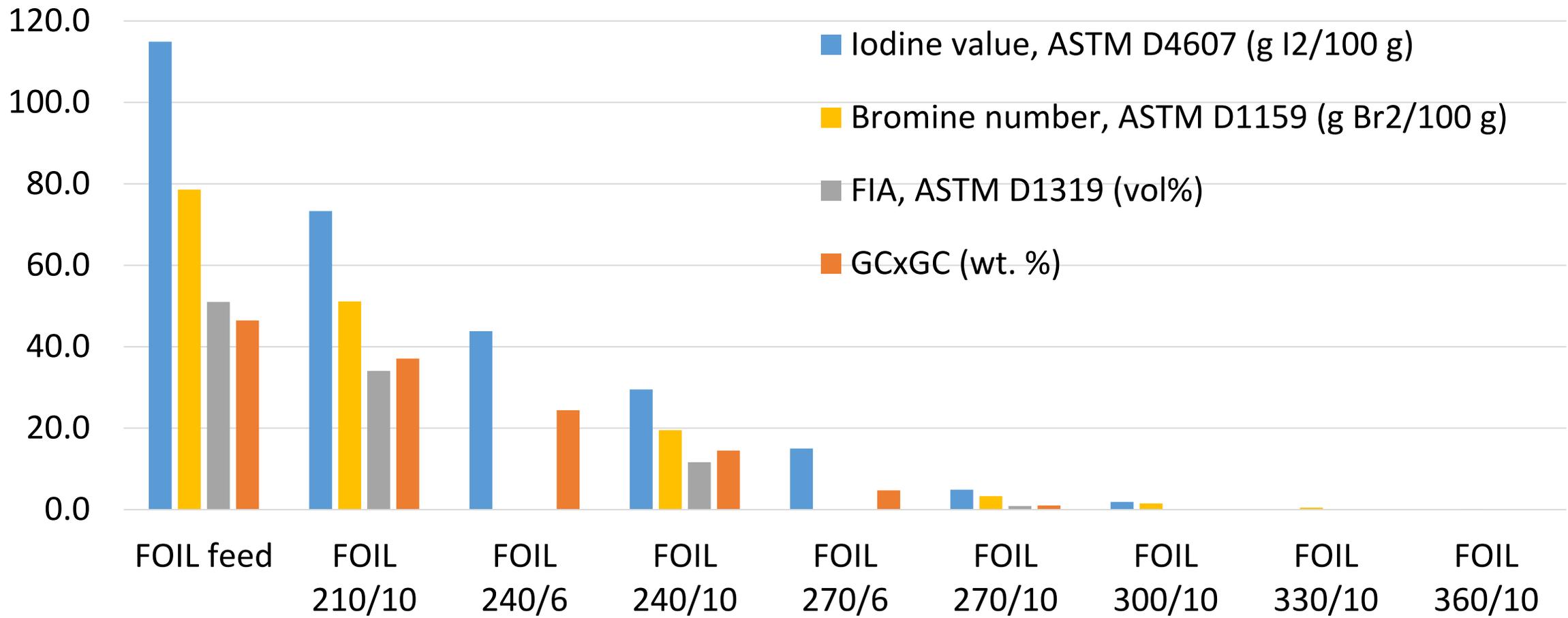
Results (totals)

Before	wt. %
<i>n</i> -alkanes	1.54
Iso-alkanes + Iso-alkenes	0.86
Monocycloalkanes + Olefins with 1 Double Bond + Linear-alkenes	4.43
Dicycloalkanes + Olefins with 2 Double Bonds	27.01
Tricycloalkanes + Olefins with 3 Double Bonds	4.60
Aromatics	53.66
Light Hydrocarbons	7.90

After	wt. %
<i>n</i> -alkanes	1.54
Iso-alkanes	0.64
Iso-alkenes	0.22
Monocycloalkanes	0.96
Olefins with 1 Double Bond + Linear alkenes	3.47
Dicycloalkanes	1.80
Olefins with 2 Double Bonds	25.20
Tricycloalkanes	1.10
Olefins with 3 Double Bonds	3.50
Aromatics	53.66
Light Hydrocarbons	7.90

Validation

comparison to other ASTM methods



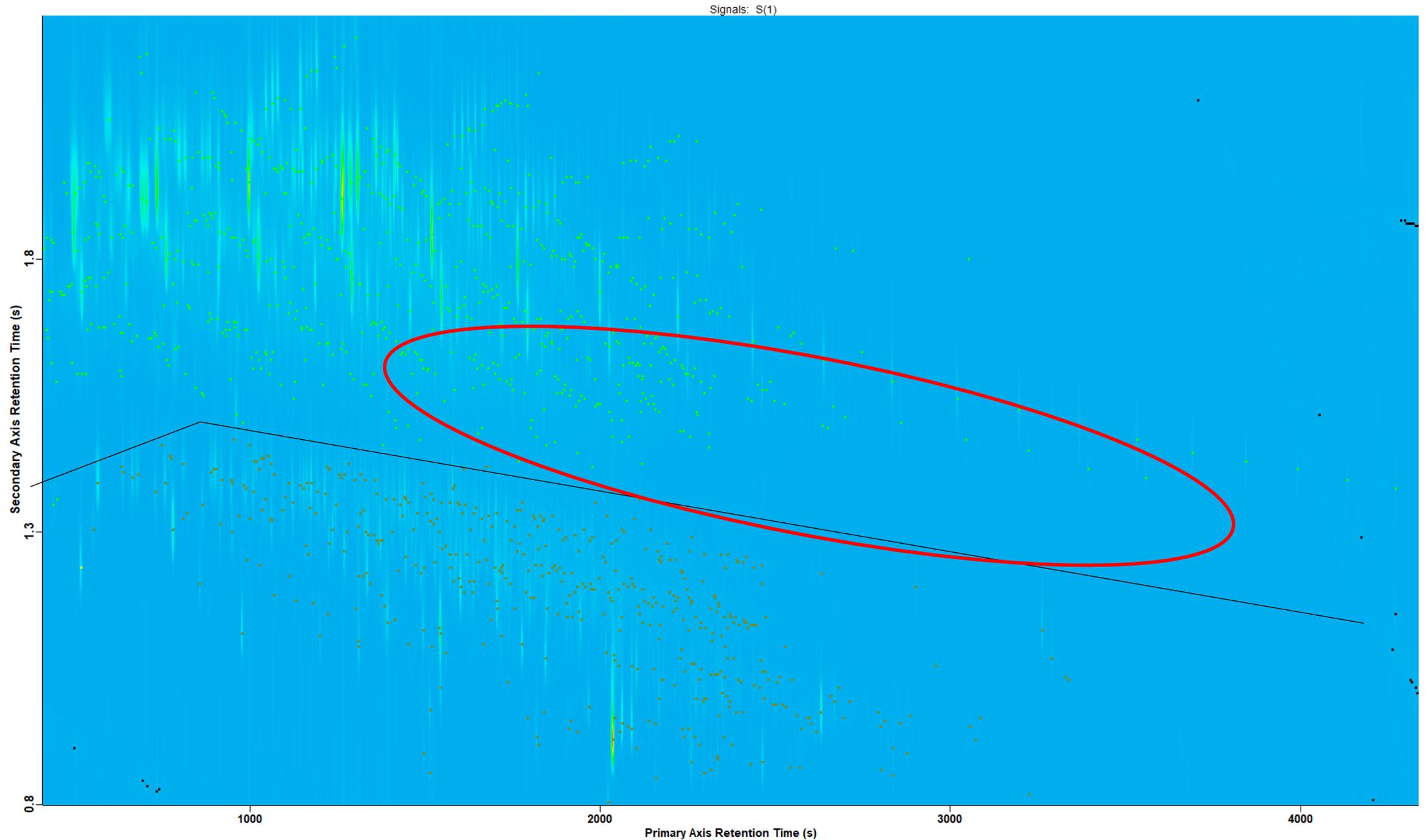
nine (9) gasoline-like samples

Methods - disadvantages

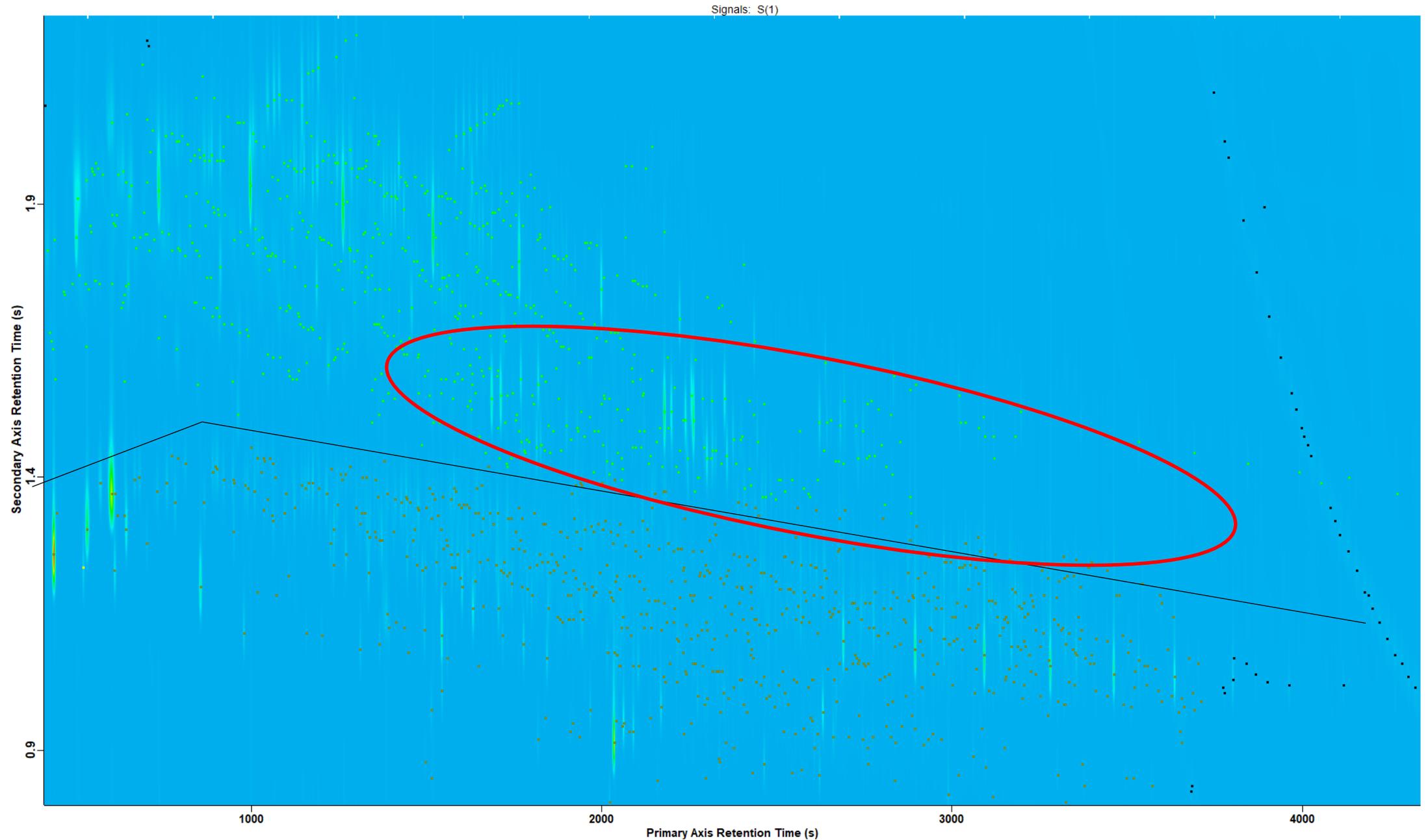
- Iodine – hard to buy in some places
- DMDS –
- Main problem – heavier isoalkenes → not shifting to aromatic region



Chromatogram – diesel-like sample pre-derivatization



Chromatogram – diesel-like sample post-derivatization



- Validate the results using GC×GC-TOFMS



Pegasus BT 4D GC×GC-TOFMS
Benchtop GC-MS with high-performance
GC×GC modulation

with thermal desorption/pyrolysis unit
with a LN2 trap

Future work

- Validate the results using GC×GC-TOFMS
- Attempt to use ChromaTOF Tile
(or beg Dr. Synovec group for their help)
- Use different derivatization methods (e.g., oxidative derivatizations, such as ozonolysis and single oxygen)

The logo for ChromaTOF|Tile. The word "Chroma" is in a dark blue font. The "TOF" part is stylized with a rainbow gradient: "T" is blue, "O" is yellow, and "F" is green. A vertical bar separates "TOF" from "Tile", which is in a dark blue font. A registered trademark symbol (®) is located between the bar and "Tile".

ChromaTOF|Tile

My awesome students...





Pegasus BT GC-TOFMS
Benchtop GC Time-of-Flight
Mass Spectrometer



QuadJet SD
GC×GC modulation system with
Flame Ionization Detection

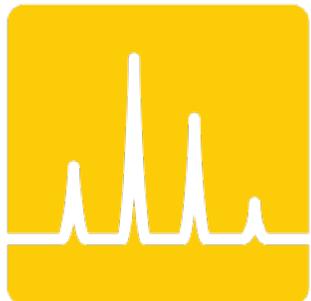


Pegasus BT 4D GC×GC-TOFMS
Benchtop GC-MS with high-
performance GC×GC modulation

What I did:

- asked for free (broken/used) stuff from the vendors
- redesigned the curriculum by including MDC instruments
- let students practice on real instruments/parts





C³AL

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Thank you, thank you....!



Dr. Pierre-Hugues Stefanuto



Dr. Katelynn A. Perrault



Dr. Dwight Stoll

**14TH Multidimensional
Chromatography
Workshop**