

Fast determination of inorganic cations and low mass amines in freshly prepared and moldy green tea samples using IC-MS

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Keywords

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Introduction

Reliable analytical methods for food and beverage samples are required in order to report ingredients for food labeling requirements, to maintain product quality, and to ensure the absence of food contamination and spoilage. Food and beverage samples are almost invariably one of the most complex sample matrices and, therefore, are among the most challenging samples to analyze.

Ion chromatography (IC), using eluent generation and suppressed conductivity detection, provides chromatographic selectivity, low chemical noise, and high compatibility with a mass spectrometer (MS).^{1,2} Additionally, the analytes leave the IC system as ions, further increasing MS compatibility.

Mass spectrometry coupled to IC provides higher selectivity and often, better detection limits. In addition, MS is one of the most universal detectors, providing powerful screening, structural, and confirmatory information.² The heated electrospray ionization (HESI), with applied high temperature aiding voltage, transforms the aqueous IC stream into a fine spray that can enter the MS detector.¹

In this application brief, separations of four inorganic cations, monoethanolamine, ethylamine, and diethylamine were achieved within six minutes in a freshly prepared green tea sample and in a three-day-old green tea sample with visible mold. The analytes were detected serially, by suppressed conductivity and single quadrupole mass spectrometry in full scan (FS) and SIM (select ion monitoring) modes without the aid of a desolvation agent during separation or ionization.

This application brief is an update to Application Note 269.³ Here, the application is performed using a Thermo Scientific™ Dionex™ Integrion™ HPIC™ system coupled to a Thermo Scientific™ ISQ™ EC single quadrupole mass spectrometer.

Experimental Ion Chromatography

- Dionex Integrion HPIC system, RFIC™ model with a second six-port high-pressure divert valve and conductivity detector
- Thermo Scientific™ Dionex™ AS-AP autosampler
- Thermo Scientific™ Dionex™ AXP-MS auxiliary pump to supply water for the suppressor

Mass Spectrometry

- ISQ EC single quadrupole mass spectrometer
- Thermo Scientific syringe pump for method optimization
- HESI II probe

Software

Thermo Scientific™ Chromeleon™ Chromatography Data System (CDS) software, 7.2 SR 6

Methods

Columns:	Thermo Scientific™ Dionex™ IonPac™ CG12A-5µm, CS12A-5µm, 3 mm i.d.	
Eluent:	33 mM Methanesulfonic acid (MSA)	
Eluent Source:	Thermo Scientific™ Dionex™ EGC 500 MSA cartridge, Thermo Scientific™ Dionex™ CR-CTC 600 trap column, high pressure degasser module	
Flow Rate:	0.5 mL/min	
Injection Volume:	100 µL	
Detection 1:	Suppressed conductivity, Dionex CERS 500e suppressor, external water mode at 0.7 mL/min by the AXP-MS auxiliary pump	
Typical Conductance Background:	< 1 µS-min	
MS Detection:*	+ESI, +3000 V, Full Scan, 18-250 <i>m/z</i> and SIM, HESI II	
Temperatures:	Vaporizer: 250 °C; Ion Transfer: 300 °C	
Flow (N ₂):	Sheath: 60 psi, Aux: 26 psi; Sweep: 0.5 psi	
Desolvation agent:	None	
SIM mode:	Ion <i>m/z</i>	CID (V)
Sodium as Na•2H ₂ O	59	10
Ammonium as NH ₄ •H ₂ O	36	2
Potassium	39	45
Magnesium as Mg ₂ •H ₂ O	66	5
Calcium	40	45
Dimethylamine	46	10
Ethylamine	46	45
Monoethanolamine	62	15
Diethylamine	74	15
Triethanolamine	150	25

*Note: The optimum HESI ion source settings and responses may vary between instruments.

Results

Figure 1 shows the IC-MS results of inorganic cations and amines in diluted tea samples — one freshly prepared and the other three days old with visible mold. The ions were detected in SIM mode from m/z 39 to 150 as native ions or water-adducts. The IC and SIM peaks are symmetrical and have strong responses, E2 to E5 ion counts (in SIM mode). Bare, unsolvated calcium m/z 40 had the lowest response at below 100 counts (e2). Notice that monoethanolamine m/z 62 (peak 2) and ethylamine m/z 46 (peak 3) had higher responses in the three-day-old tea than in the freshly prepared tea sample. Small changes in the conductivity detection signals were measured between the two samples. However, peaks 2, 3, and 6 were measured at the limits of detection and peaks 2 and 3 were not resolved using suppressed conductivity detection. It is worthwhile to

note that MS provided improved resolution between peaks 2 and 3 and a higher signal-to-noise ratio, which allowed a straightforward identification of moldy tea samples based on peaks 2, 3, and 6. More information can be found in the Thermo Scientific™ AppLabs Library of Analytical Applications.⁴

References

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- Thermo Scientific Application Note 269: Identification and Quantification at ppb Levels of Common Cations and Amines by IC-MS. <https://apps.lab.thermofisher.com/Search?SearchText=an269>
- Thermo Scientific AppLabs Library of Analytical Applications. <https://apps.lab.thermofisher.com>

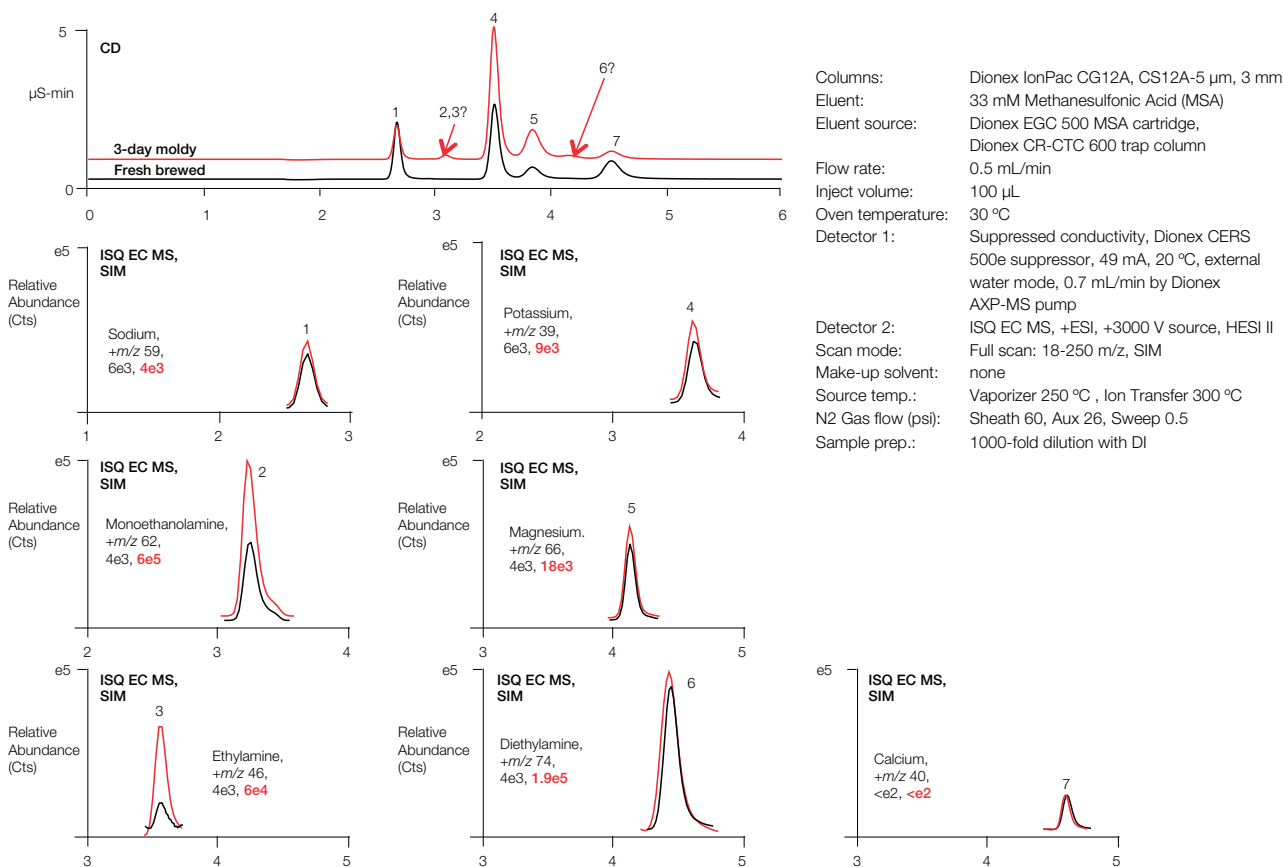


Figure 1. Determination of inorganic cations and amines in green tea samples A) freshly prepared and B) three days old with visible mold.

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